



***CAPE and Thermodynamic Property Packages:
Air Liquide approach***

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CAPE and Thermodynamic Property Packages: Air Liquide approach

- **History of thermodynamic modelling and process simulation tools at Air Liquide**
- **Evolution of Air Liquide business since the 90s**
- **Current difficulties**
- **Conclusions and future steps**

■ 1960 – 1988: In-house development

✓ Thermodynamic properties of pure substances

- 1976: Publication of Gas Encyclopaedia (Elsevier - ISBN 0-444-41492-4 -2nd reprint 1992, 3rd reprint 2002)
 - *138 monographs of gases (physical, thermodynamic and transport properties – flammability - biological properties - precautions in handling and storage - leak detection and analysis – material compatibility)*
 - $N_2, O_2, Ar, CO_2, H_2, CH_4, C_2H_4, C_2H_2, C_3H_8, NH_3$
 - Detailed P-T tables (phases equilibrium, vapor pressure, density, compressibility factor, enthalpy, entropy, heat capacity, viscosity, thermal conductivity)
 - Two types of equation used to generate data for a given substance: empirical correlation dedicated to each property or PVT equation of state (virial development) calibrated on measured values from literature
 - High accuracy (generally lower than 0.1%) due to large number of experimental points on which calibration is done

✓ Thermodynamic properties of mixtures

- 1970 -1972: Development of thermodynamic equation of state dedicated to air cryogenic distillation (N_2, O_2, Ar)
 - *Empirical equation of Benedict-Webb-Rubin [1], mixing rules of Starling [2]*
 - *Regression of pure substance parameters and binary interaction coefficients on in-house experimental values*
- 1972 -1975: Development of thermodynamic equation of state dedicated to natural gas liquefaction and CO separation and purification by methane washing
 - *Modification of Redlich-Kwong equation of state [3] to reproduce more accurately liquid phase properties along the saturation curve*
 - *Regression of binary interaction coefficients on experimental values from literature*
 - *Maximum number of constituents: 20 among 39*

[1] Benedict M, Webb GB, Rubin LC, *J. Chem. Phys.*, 1949, 8, 334-344.

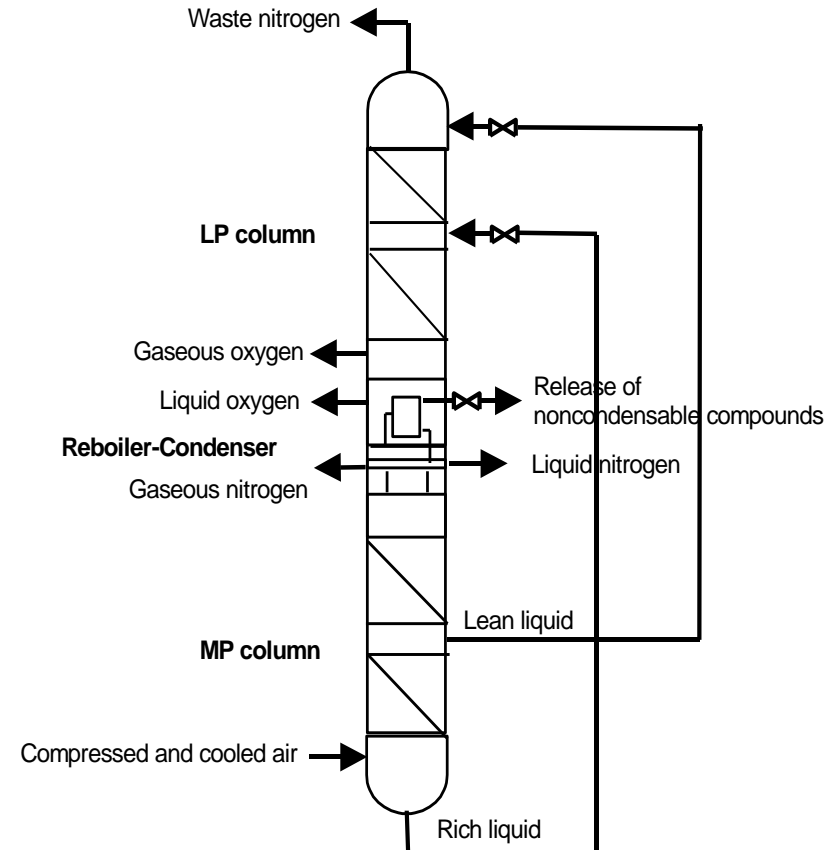
[2] Starling KE, *Hydro. Process.*, 1971, 101-104.

[3] Redlich O, Kwong JNS (1949) *Chem. Rev.*, 44, 233-244.

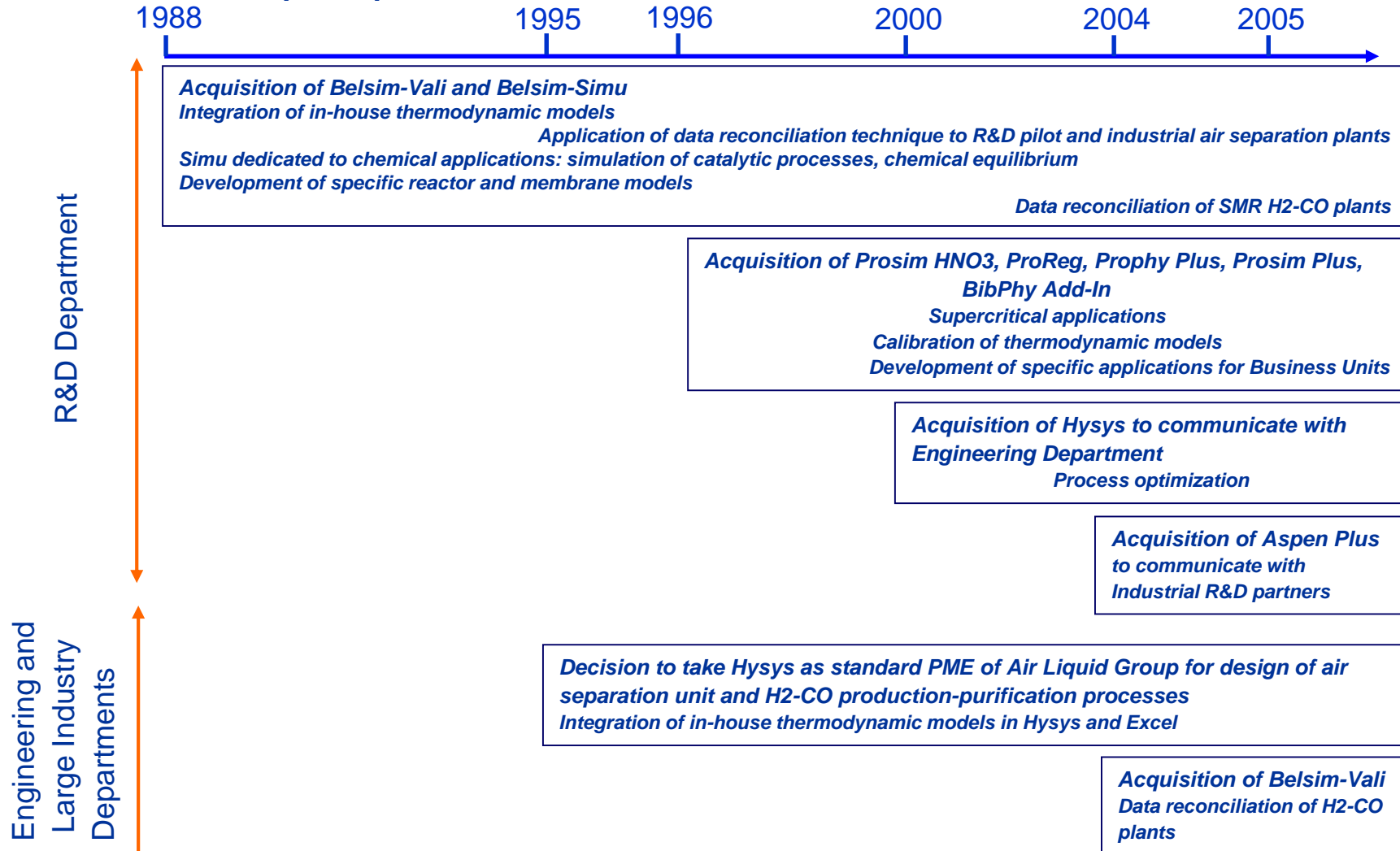
■ 1960 – 1988: In-house development

✓ Process simulation tools

- Development of model dedicated to the simulation of conventional double column apparatus for air distillation
 - *Brazed aluminum plate-and-fin heat exchanger*
 - *Compressor, pump, expander*
- Development of absorption column and stripper models for CO separation application



■ 1988 – 2004: Acquisition and use of commercial Process Modelling Environment (PME)



■ Overview of the current situation

- ✓ Cohabitation of four PMEs: Hysys, Belsim, Prosim, Aspen
- ✓ Different versions of in-house thermodynamic models
- ✓ Specific Process Modelling Components (PMC) compliant with only one PME
- ✓ Continuation of in-house development: Matlab®, Excel®, Fortran
 - Development of specific PMCs by R&D and transfer to Engineering Department or Business Unit (Centralized production facilities and Cylinder filling plants)
 - Development through academic collaboration (PhD, ...)

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■ Diversification

- ✓ Strong increase of H₂-CO business
 - Modelling, design and operation of H₂-CO production plants by steam reforming of natural gas
 - *Reactors: pre-reformer, reformer, water-gas shift reactor*
 - *Separation units: water condensation from syngas, CO₂ removal (amine washing), CO separation (methane washing)*
 - Hydrogen purification by membrane or PSA: dew point of HC-H₂S-H₂ mixtures
- ✓ New projects linked to CO₂ capture, purification and sequestration
 - CO₂ separation from oxy-combustion gases or blast furnace by PSA or distillation
- ✓ Cylinders: a more and more accurate demand
 - Deregulation of natural gas distribution
 - *Increase of the demand of natural gas mixture used for analyzer calibration*
 - *Recommendation to supply complete phase envelop of the mixture (dew and bubble curves, critical point location)*
 - Liquid mixture for petrochemical industry
 - *Prediction of liquid phase composition evolution during cylinder blow-off*

■ Health and Electronics

- ✓ High purity (ppm-ppb) and high accurate mixtures (0.1%)

■ Historical core business

- ✓ Increase of air separation plant capacity: oxygen production > 4 000 t/d
 - Safety: impurities accumulation in air separation units
 - *Formation of solid CO₂ and N₂O in liquid oxygen*
 - *HC solubility in liquid oxygen*

→ A more and more complex and diversified demand for thermodynamics and simulation tools

✓ Thermodynamic properties

- Nature of the substances
- Nature of the mixture
 - *Water-Liquid-Vapor Equilibrium*
 - *Liquid-Liquid-Vapor Equilibrium*
 - *Solid-Liquid-Vapor Equilibrium*
 - *Electrolyte solutions (acid or basic)*
- Robustness and accuracy of LV equilibrium resolution algorithm
 - *Location of critical point of mixtures*
 - *Convergence around cricondenthem and cricondenbar points*

✓ Process Modelling Components

- Detailed reactor model for steam methane reforming
 - *Kinetic model: gas-solid reaction, diffusional limitations, carbon formation*
 - *Heat transfer by radiation, convection and conduction*
- Pressure Swing Adsorption model
 - *Time-dependent and cyclic process*
 - *Kinetic of adsorption-desorption*

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■ General limitations

- ✓ Precarious communication and model transfer between the different CAPE actors
 - R&D and Advanced Technologies Department
 - Engineering Department
 - Business Units
- ✓ Multifaceted contexts
 - R&D: exploratory and development projects, feasibility study, technical support
 - Engineering projects (design and quotation)
 - Industrial plant operation and control

■ Multiplicity of PME

- ✓ Different results with “same thermodynamic model” from different PMEs
 - Origin of differences
 - *Database of pure substance properties*
 - *Version of the equation of state (alpha function) or activity coefficient model*
 - *Mixing rules*
 - *Resolution algorithms of LV equilibrium*
 - *Reference state of enthalpy calculation*
 - *Model or correlation used for the calculation of liquid molar volume and liquid fugacity in standard state*
- ✓ No interoperability between PMEs
 - Standard CMPs of one PME not compliant with the other PMEs
 - CMPs (reactors, membranes) developed by AL in one PME not compliant with the other PMEs

■ In-house tools

- ✓ Gas Encyclopaedia: e-version available on Air Liquide web site
 - P-T tables of thermodynamic properties exportable in Excel sheet
 - Temperature and pressure interpolations are required
 - ✓ In-house simulation models (Matlab®, Excel®, Fortran)
 - Limitations due to isolated use
 - *No possibility of coupling CMPs coming from PME's or another model*
 - *Requirement: (re-) programming of pure substance properties and equation of state*
 - *Approximate programming of pure substance properties or use of correlations outside their validity range can lead to propagation of errors*
 - Management of versions
 - ✓ Thermodynamic models
 - Method of integration in Excel® and in the PME's Hysys® and Belsim-Vali®
 - *Fraction coming from the original model*
 - *Fraction already programmed in the PME (pure compound properties, solving method, ...)*
 - *Different results in different environments*
 - Not available in all environments (PME's, Excel®, Matlab®, ...)
 - Many developers, many users ... lead to many versions...even in the same environment
 - Traceability and justification of modifications (mixing rules, binary interaction coefficient calibration, addition of pure components, pure substance parameters)
 - *Increase the distance from initial objective*
 - *Fuzzy validity range*
- ➔ Difficulties of perpetuation of in-house knowledge and of integration in commercial tools

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■ Main actions to improve the existing situation

- ✓ Adoption of a thermodynamic standard usable at each step of process development
 - Database of pure substances properties
 - Database of thermodynamic models and associated resolution algorithms
- ✓ Development of concept of CO “thermodynamic property package” dedicated to given applications
 - Interoperability of packages: PMEs, Excel®, Matlab®
 - Consistency of results throughout different applications
 - Perpetuation of in-house knowledge: database of packages
 - *Detailed description and validity range*
 - *References of experimental values (literature, in-house)*
 - Improvement of accessible information quality for a relevant later re-use
- ✓ Adaptation of existing simulation models to CO standards and new developments CO compliant
 - Compliant with CO thermodynamic property packages
 - Interoperability: PMEs

■ September 2004 – November 2005

- ✓ Test of Simulis® Thermodynamics (Prosim SA) in R&D
 - Development of a thermodynamic package dedicated to CO₂-CO-N₂-H₂ mixtures
 - *SRK equation of state*
 - *Alpha function of Boston-Mathias*
 - *Calibration of binary interaction coefficients on measurements from literature*
 - *Validation of results on ternary experimental data*
 - Creation of CO property package
 - *Plug tests in Hysys 2004 with the help of Prosim SA, CO-LaN and Hysys hotline*
 - *November 2005: final validation of the package integrability in Hysys 2004 environment*
 - *Deployment of the package in R&D department*

■ January 2006: Decisions

- ✓ Air Liquide R&D thermodynamic standard: Simulis® Thermodynamics associated to DIPPR® database of pure substances
- ✓ Test of Simulis® Thermodynamics by Engineering Department

■ Future steps

- ✓ Test of Simulis® packages in other environments
 - Belsim-Vali®
 - Matlab®
 - Aspen Plus®
- ✓ Creation of a Simulis® packages database
- ✓ Test of an in-house simulation model adaptation to CO standard
- ✓ Deployment of package development methodology
 - Bibliographic study (measurements, modelling)
 - Choice of a thermodynamic profile
 - Evaluation of calibration requirement, if yes
 - Requirement evaluation of additional experimental values in studied conditions
 - *Question: what is the minimum number of supplementary experimental values to be measured to avoid significant errors*
 - Campaigns of measurements
 - Calibration of the model
 - Creation of the package
 - Integration in the package database



Thank you for your attention

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