

Thermo SIG Progress Report 2013

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Thermodynamics Special Interest Group (Thermo SIG)

Task:

Develop, maintain and promote Thermodynamic and Physical Properties interface specifications

Key Responsibilities:

- ◆ **Maintain and manage existing interface specifications**
- ◆ **Assess expansions of interface specifications**
- ◆ **Manage the development of expansions**
- ◆ **Help organizations to develop implementations**

Thermo SIG Members, September 2013

- ◆ Alan Scott, contractor for TÜV-SÜD-NEL
- ◆ Andrew Lintern, HTRI
- ◆ Bjorn Maribo-Mogensen, Technical University of Denmark
- ◆ Ensheng Zhao, Honeywell
- ◆ Jasper van Baten, AmsterCHEM (SIG co-leader)
- ◆ Michel Pons, CO-LaN
- ◆ Murugesh Palanisamy, Honeywell
- ◆ Paul Zhou, Honeywell
- ◆ Rafael Lugo, IFP
- ◆ Richard Szczepanski, Infochem Computer Services
- ◆ Sergej Blagov, BASF (SIG co-leader)
- ◆ Suphat Watanisiri, AspenTech
- ◆ Xiaozheng-Sara Wang, Honeywell

Chemical Reactions Interface (September 2012)

- ◆ **Several issues exist with current v1.0 Reactions specification**
 - Reaction basis is not clearly defined
 - Units of measure require revision (non-SI)
 - Several concepts not well defined
- ◆ **Large overlap with v1.1 Thermodynamic and Physical Properties**
 - Compound definitions
 - Material contexts
- ◆ **Conclusions:**
 - Need for a new **Reactions** interface specification
 - Need for integration with v1.1 **Thermodynamic and Physical Properties** interface specification
- ◆ **Status:**
 - **Revision done; RFC starts in October/November, 2012**

Chemical Reactions Interface: Key Features (September 2012)

◆ Generality in focus

- Different phases might be specified for reactants and products on component basis
- Various reaction types
 - Single phase reactions
 - Interfacial and Surface reactions
 - Homogeneous and Heterogeneous reactions
- Formulation on the true species basis

◆ Very compact

- Only 4 reaction properties supported
 - Reaction Rate
 - Chemical Equilibrium Deviation function
 - Chemical Equilibrium Deviation tolerance (constant value)
 - Enthalpy of Reaction

Finalizing of Chemical Reactions Interface

October / November, 2012

- **Preparation of the finalized version of Chemical Reaction Interface started (J. van Baten, M. Pons, S. Blagov)**
- **Major drawbacks recognized (mainly feedback from M. Pons):**
 - True species approach does not allow for hiding any details of reaction mechanism
 - Stoichiometry must always be explicitly defined and exposed
 - No means provided for handling systems of special type, e.g. polymers, electrolytes, etc.
 - Reactive phase equilibrium calculator not defined, which is, however, of most interest for users
- **Major revision required**

Chemical Reactions Interface Specification. I

Comparison of revised 1.1 Reactions interface to existing 1.0

- ◆ Integration with the CAPE-OPEN version 1.1 thermodynamic standard, as opposed to 'stand-alone' interface
- ◆ This eliminates the redundant definitions for compounds
- ◆ Units of measure are SI (as opposed to version 1.0)
- ◆ Reactions can be served by Property Package, or by Reaction Package

Chemical Reactions Interface Specification. II

Comparison of revised 1.1 Reactions interface to existing 1.0

◆ Reaction Object discarded, this task is to be performed by the Material Object

◆ Reaction Context discarded

- for Reaction servers this is the Material Context
- a Unit Operation can obtain reactions ONLY from an associated Material Object (e.g. connected stream)

Chemical Reactions Interface Specification. III

Comparison of revised 1.1 Reactions interface to existing 1.0

- ◆ *Equilibrium constant* discarded in favour of *EquilibriumDeviation*
- ◆ Reaction phase replaced by vector: one phase for each compound (e.g. solid C can react with gaseous O₂)
- ◆ Reaction rate basis (reaction domain): *Phase*, *PhaseInterface*, *CatalystMass*, *Surface*
- ◆ Both *Reaction Rate* (stoichiometry required) and *CompoundReactionRate* (stoichiometry is optional) are supported

Chemical Reactions Interface Specification. IV

New in version 1.1 specification

- ◆ Reaction attributes: *Mechanism, Kinetic, Equilibrium, Comment, Catalyst, FullConversion, etc.*
- ◆ Equilibrium Reaction Sets: sets of equilibrium reactions that need to be considered simultaneously
- ◆ Overall pseudo-kinetic reaction: several kinetic reactions lumped into one, for which only per-compound reaction rates are provided
- ◆ Reaction Groups: optional arrangement of related reactions, for aiding reaction selections in private GUIs
- ◆ Reaction Alternatives: multiple versions of the 'same' reaction with different properties, e.g., rate expression

Chemical Reactions Interface Specification. V

Reactive Phase Equilibrium Calculator

- ◆ A reaction for which the overall equilibrium composition is part of the result
- ◆ No Reaction Package needed
- ◆ Performed by Equilibrium Server (e.g. Property Package or native PME thermo)
- ◆ Requires new interface: *ICapeThermoReactiveEquilibrium*
- ◆ *True* composition is separately stored on Material Object
- ◆ *MolarReactionRatio* - an additional output argument:

$$= \frac{\text{Total mole number AFTER}}{\text{Total mole number BEFORE}}$$

Chemical Reactions Interface Specification. VI

True vs Apparent Compounds

- ◆ *ICapeThermoCompounds* exposes union of *True* and *Apparent* compounds
- ◆ Compounds can be *True*, *Apparent* or both
- ◆ Associated equilibrium calculation is reactive
- ◆ Equilibrium calculation operates with *True* compounds only
- ◆ State of Material Object is defined by pressure, temperature and *True* composition
- ◆ Conversion between apparent composition and *True* composition done by Property Package (if provided)
- ◆ Support for *Apparent* properties (e.g. fugacity)

Summary

- ◆ Thermo v1.0 and 1.1 specifications maintenance
 - Sustainable usage
 - Very few issues for clarification reported
 - v1.0: CalcType for P & T: both, 'Mixture' and UNDEFINED are to be allowed
- ◆ In focus: Chemical Reaction Interface Specification
 - Major revision of draft document was required
 - Ongoing work on a new draft
(weekly phone meetings: J. van Baten, M. Pons, S. Blagov)
 - Deadline still not fixed
(for some concepts there is no standard solution and they need further elaboration)

Questions?

Thank you for your attention!



Go CAPE-OPEN!